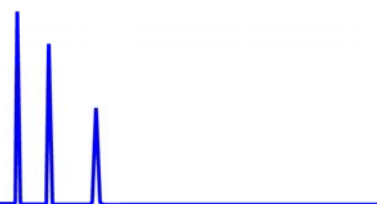




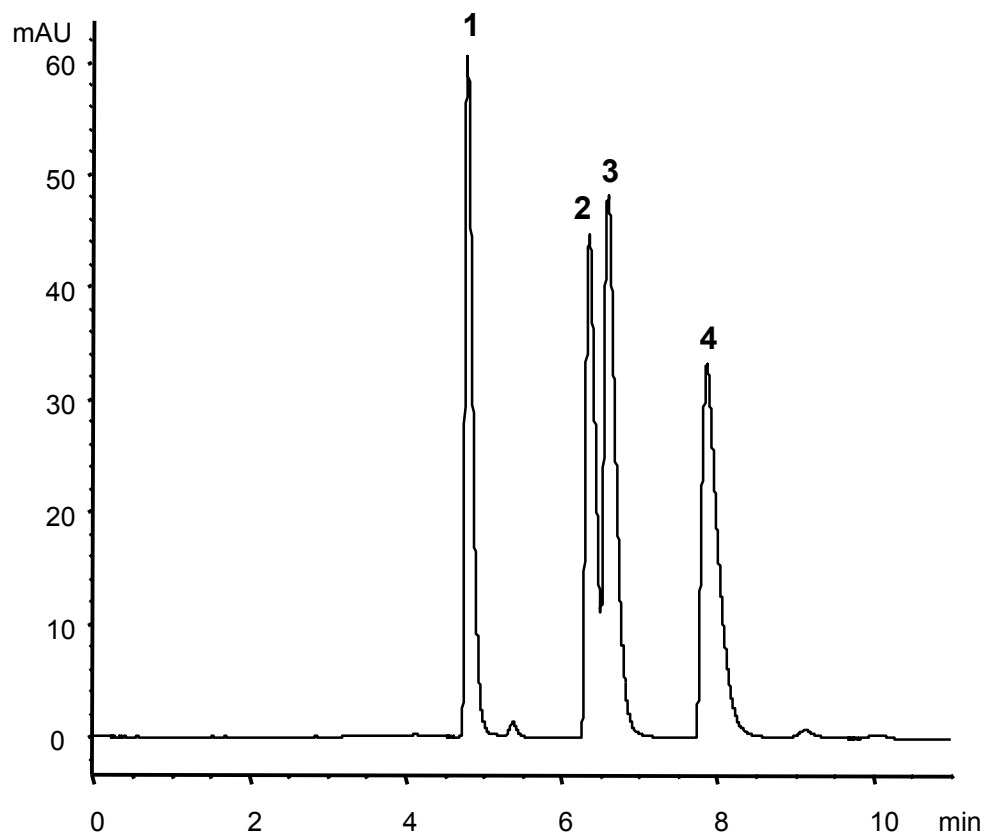
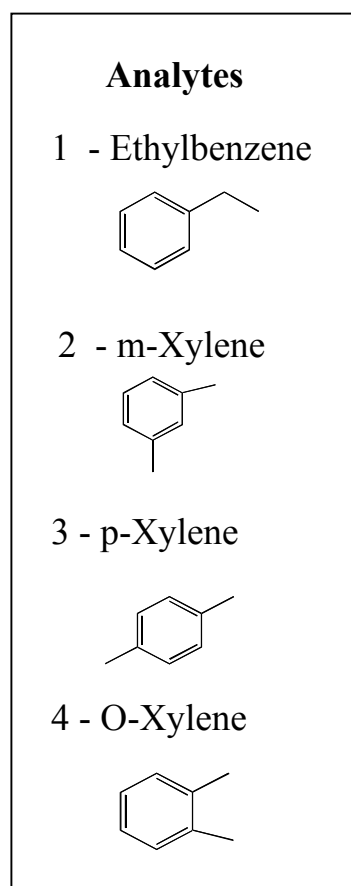
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# Technical Bulletin #134



... For Peak Performance

## Isomer Separation on ZirChrom®-CARB



### LC Conditions

Column: A: ZirChrom®-CARB, 150 × 4.6 mm;

Mobile Phase: 50/50 A/B

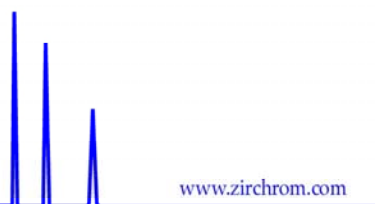
A: Acetonitrile

B: Water

Flow rate: 1.0 mL/min.

Temperature: 30 °C

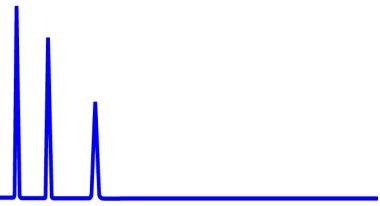
Detection: 254 nm





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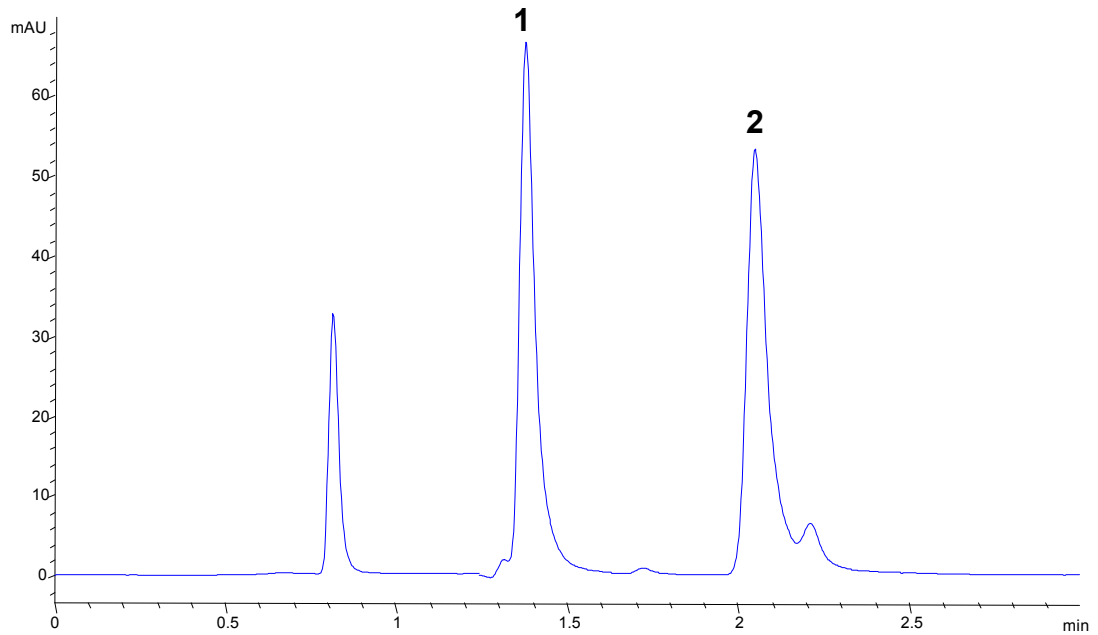
Technical  
Bulletin #203



... For Peak Performance

**Cosmetics on ZirChrom®-CARB**

**Analytes**  
1 - Allantoin  
2 - Bronopol



**LC Conditions**

Column: ZirChrom®-CARB, 100 × 4.6 mm

Mobile Phase: 20/80 A/B

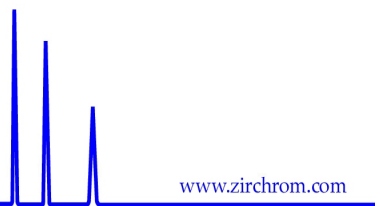
A: Acetonitrile

B: Water

Flow rate: 1.0 mL/min.

Temperature: 25 °C

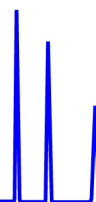
Detection: 200 nm





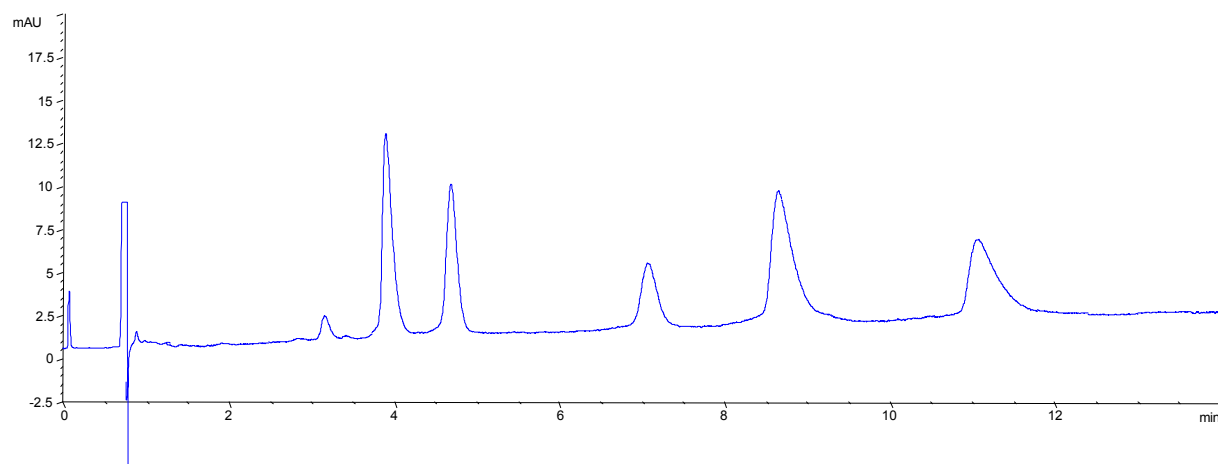
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# Technical Bulletin #244



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## Separation of a Steroid and its Metabolites on ZirChrom®-CARB



Compound	Tr	Area	Height	k'	a [n/(n-1)]	Tailing Factor	Calc. Plates	Calc. Rs
A	3.371	14.1	2.6	1.86	N/A	1.47	8736	N/A
B	4.505	87.2	8.6	2.83	1.52	2.38	4463	4.20
C	5.247	72.6	8.8	3.46	1.22	1.54	9145	3.38
D	7.661	55.8	4.7	5.51	1.59	1.54	9414	7.64
E	9.955	125.9	6.2	7.46	1.35	1.92	5433	4.25
F	11.542	75.5	3.7	8.81	1.18	1.72	7233	2.92

### LC Conditions

Column: ZirChrom®-CARB, 100 x 4.6 mm

Mobile Phase: 20/30/30/20 A/B/C/D

A: ACN

B: MeOH

C: THF

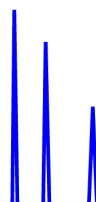
D: Water

Flow rate: 1.0 mL/min.

Temperature: 60 °C

Injection volume: 5 µL

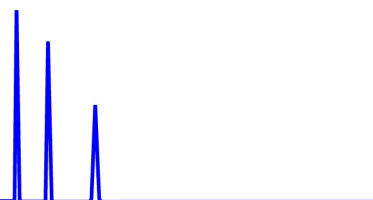
Detection: 200 nm





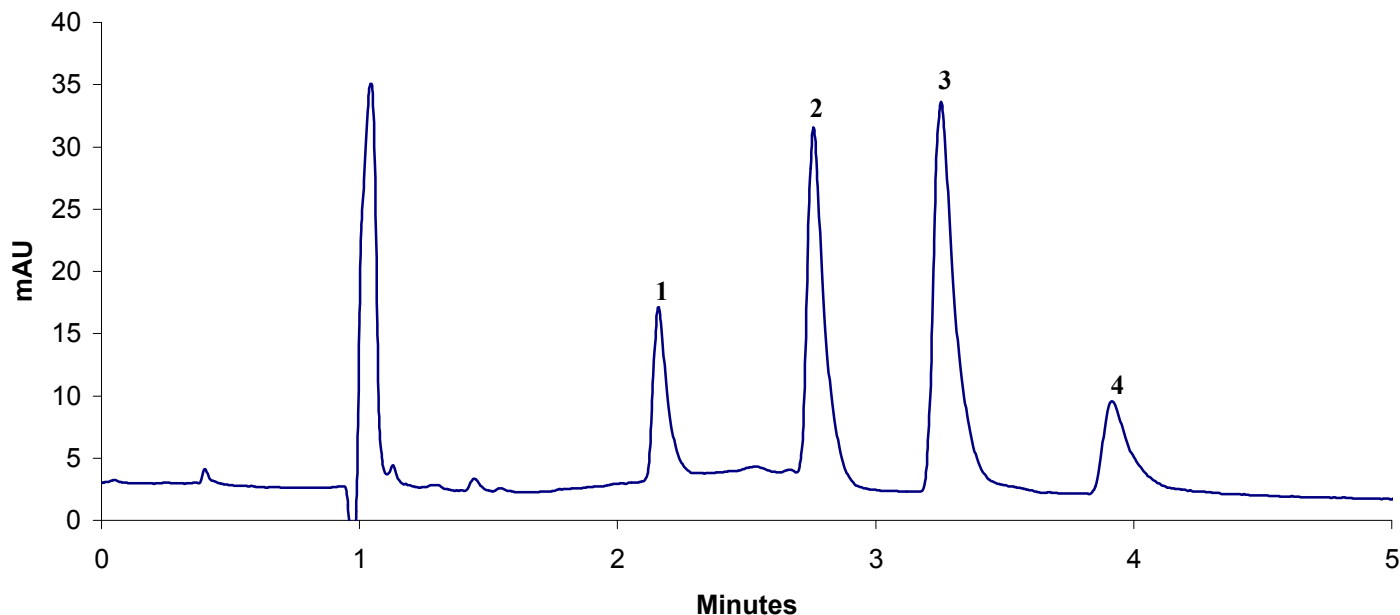
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Technical  
Bulletin #265



... For Peak Performance

## Separation of Corticosteroids on ZirChrom®-CARB



**Analytes\***

1 - Dexamethasone, 2 - Prednisone, 3 - Prednisolone, 4 - Betamethasone

\*Peak assignments revised 09-04-03

### LC Conditions

Column: ZirChrom®-CARB, 150 mm x 4.6 mm i.d.,  
(part# ZRO1-1546)

Mobile Phase: 60/10/30 A/B/C

A: ACN

B: MTBE

C: Water

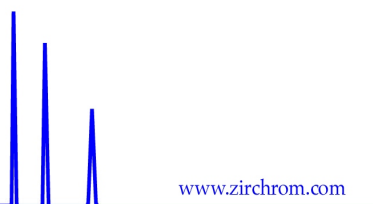
Flow rate: 1.5 mL/min.

Temperature: 80 °C

Injection volume: 15 µL

Detection: 215 nm

Pressure Drop: 160 bar





# Fast Methods for Structurally Similar Compounds, Carbon HPLC Columns

## Technical Bulletin #266

Dwight Stoll, Clayton V. McNeff, Peter W. Carr  
ZirChrom Separations, Inc.

Method development can be challenging for structurally similar compounds in RPLC. A new carbon-clad zirconia phase offers dramatically different chromatographic selectivity for these types of compounds when compared to traditional silica-based bonded phases. This note shows baseline separation of 6 structurally similar sulfate-steroid conjugates using a zirconia-based ZirChrom-CARB HPLC column.

### Introduction

Method development in reversed-phase liquid chromatography (RPLC) has traditionally been difficult for molecules which are geometric isomers or structurally very similar. In bonded-phase silicas, the partition mechanism responsible for retention in RPLC often does not offer adequate chemical selectivity for such compounds. On the other hand, carbon-based phases, provide retention in RPLC through an adsorption mechanism which often times increases the chromatographic selectivity for these types of compounds and dramatically increases the chances of resolving pairs of these analytes. In addition to this enhanced chromatographic selectivity, carbon-based phases also offer increased chemical and stability of the stationary phase.

### Carbon-clad zirconia

ZirChrom Separations, Inc. has developed new materials using zirconia as a stationary phase support, and patented chemical vapor deposition technology to produce a carbon-clad zirconia particle suitable for use in reversed-phase liquid chromatography. The surface bond is C-C, which is extremely resistant to chemical and thermal attack. The authors have run mobile phases at very high pH (1M NaOH), very low pH (0.5M HNO<sub>3</sub>), and at elevated temperature (up to 150 °C), and have not observed loss of stationary phase.

### Experimental

A mixture of sulfate-steroid conjugates was separated at elevated temperature using a ZirChrom®-CARB column. The separation conditions were as follows:

Column: 4.6 mm x 100 mm ZirChrom-CARB  
Mobile Phase: Gradient elution from 55/5/40 to 90/5/5 A/B/C from 0 to 4.5 minutes.  
A: Acetonitrile  
B: Tetrahydrofuran  
C: 25mM Ammonium fluoride, 10mM Ammonium acetate, pH 5.6  
Temperature: 80 °C  
Injection Vol.: 10 µl  
Flow rate: 3.0 ml/min.  
Pressure Drop: 195 bar  
Detection: UV at 270 nm

Figure 1 – Structures of steroid-sulfate Conjugates

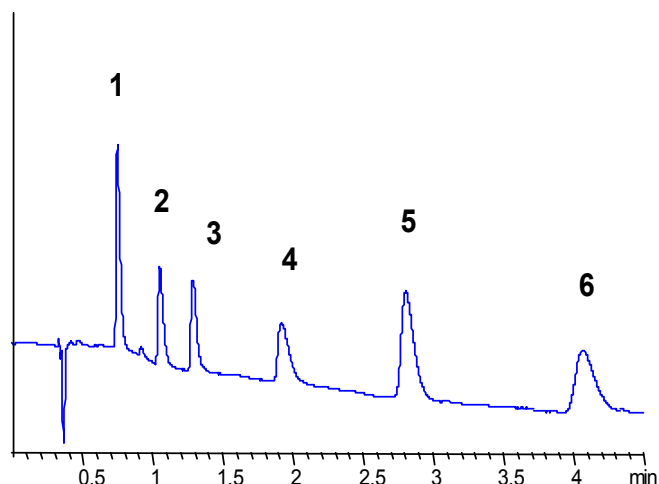
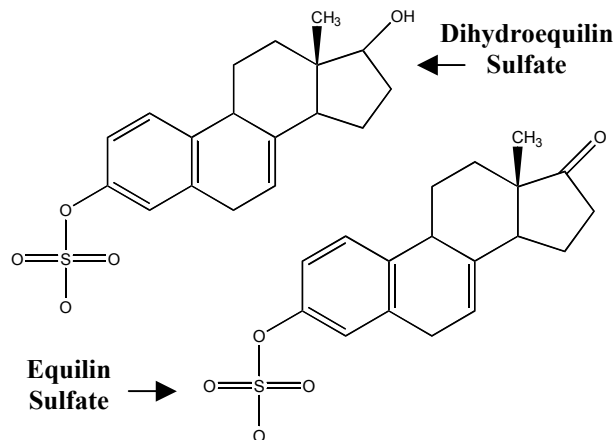


Figure 2: Steroid conjugate selectivity

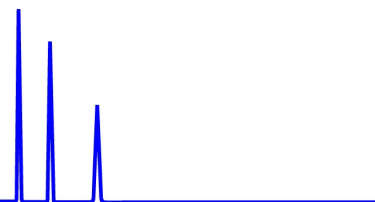
1=Dihydroestradiol sulfate, 2=Dihydroequilin sulfate, 3=Equilin sulfate, 4=Estrone sulfate, 5=Equilenin sulfate, 6=Dihydroequilenin sulfate

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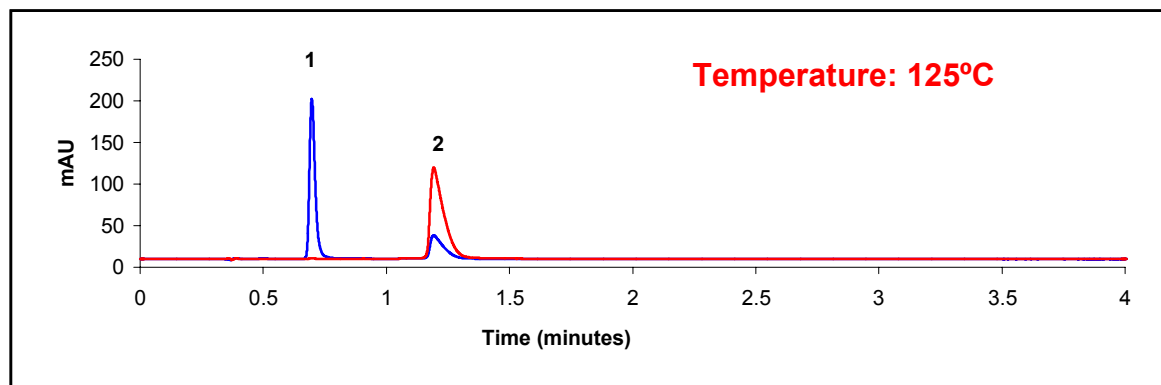
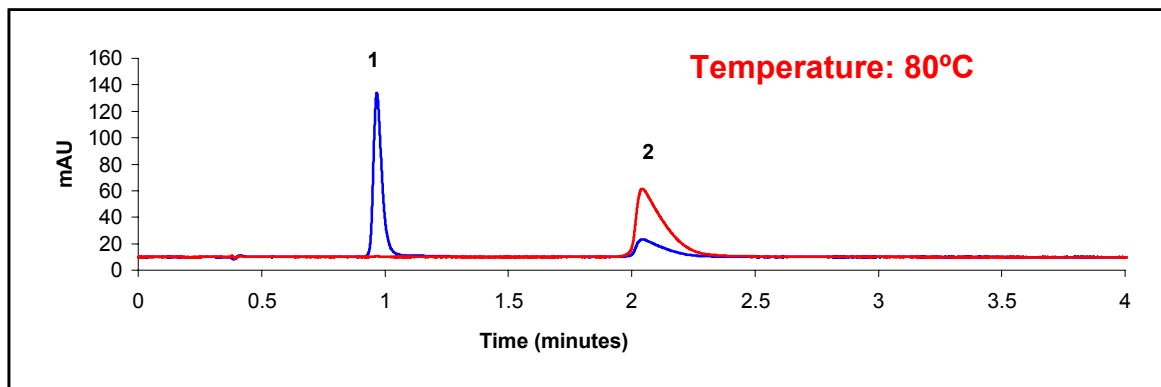


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 Technical  
 Bulletin #273


... For Peak Performance

## Separation of Reproductive Steroids on ZirChrom®-CARB



Column: ZirChrom®-CARB 50mm × 4.6 mm i.d.

Mobile Phase: Isocratic 42.5/42.5/15 A/B/C

A: Acetonitrile

B: 1-Butanol

C: Water

Detection: Blue-242nm Red-280 nm

Injection: 1 uL

Solutes: 1. Testosterone

2. Estradiol

Flow Rate: 1.5 ml/min

Temperature: 80°C and 125°C with Metalox 200-C





# Fast Separation of Eleven Nitroaromatic Compounds on ZirChrom®-CARB

Clayton McNeff, Ph.D. and Kelly Johnson  
ZirChrom Separations, Inc.

## Technical Bulletin #280

This technical bulletin details the separation of eleven closely related nitroaromatic compounds, namely RDX, HMX, Nitrobenzene, 2-Nitrotoluene, Tetryl, 2,6-Dinitrotoluene, 4-Nitrotoluene, 1,3-Dinitrobenzene, 2,4-Dinitrotoluene, 2-amino 4,6-dinitrotoluene, 1,3,5-Trinitrobenzene. Our customers have reported that similar separations on silica-based phases produce run times as long as thirty minutes. Here we report a method on ZirChrom®-CARB at a column temperature of 125°C in under 4 minutes using a Metalox™ 200-C heater.

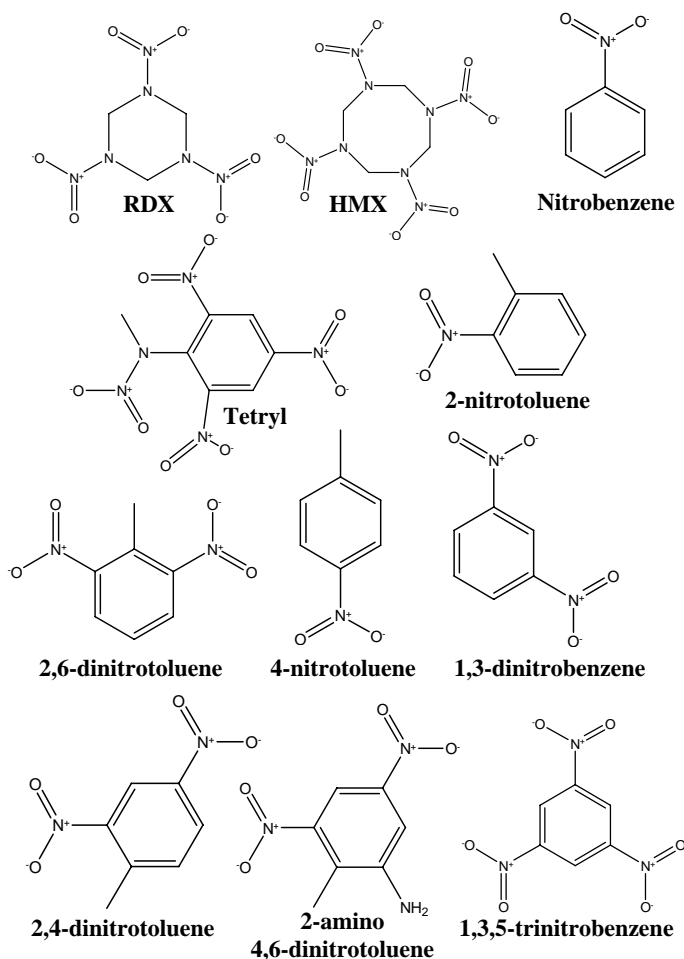


Figure 1: Structures of Explosives

### Introduction

The rapid and accurate detection of nitroaromatic compounds is difficult due to the structural similarity of the compounds (see Figure 1). A new method developed at ZirChrom employs the unique temperature stability and surface chemistry of zirconia to achieve baseline resolution of these compounds in less than 4 minutes.

### Experimental

A mixture of nitroaromatics was separated at 125°C using a ZirChrom®-CARB column (See Figure 2). The separation conditions were as follows:

Column: ZirChrom®-CARB,  
150mm x 4.6mm i.d., (part#ZR01-1546)  
Mobile phase: Isocratic Pre-mixed 35/15/50  
Acetonitrile/Tetrahydrofuran/20mM ammonium carbonate pH 5.7, 10mM octylamine  
Flow Rate: 2.0 ml/min.  
Temperature: 125°C (Metalox™ 200-C Heater)  
Detection: 254 nm  
Inj. Volume: 1µl

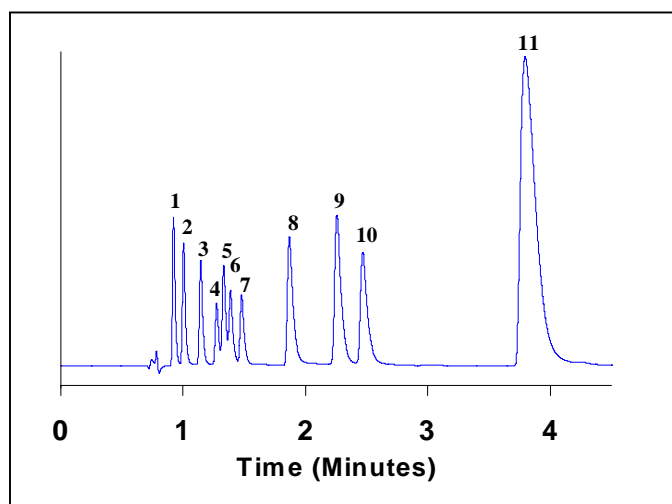


Figure 2: Separation of eleven nitroaromatics on ZirChrom®-CARB. 1=RDX, 2=HMX, 3=Nitrobenzene, 4=2-Nitrotoluene, 5=Tetryl, 6=2,6-Dinitrotoluene, 7=4-Nitrotoluene, 8=1,3-Dinitrobenzene, 9=2,4-Dinitrotoluene, 10=2-amino 4,6-dinitrotoluene, 11=1,3,5-Trinitrobenzene

This separation allows for clear identification and quantification of these compounds without the use of expensive MS detection. The separation is also completed using isocratic conditions, thus facilitating a more reproducible transfer from LC to LC.

### Acknowledgments

We would like to thank Richard Burrows, Severn Trent Laboratories, for his many helpful discussions.

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# Fast Separation of Acrylamide Monomer from Acrylic Acid

Clayton McNeff, Ph.D. and Dwight Stoll  
ZirChrom Separations, Inc.

## Technical Bulletin #281

The unique selectivity, pH and temperature stability of the ZirChrom<sup>®</sup>-CARB phase allows baseline resolution of acrylamide and acrylic acid using a 4 minute gradient method.

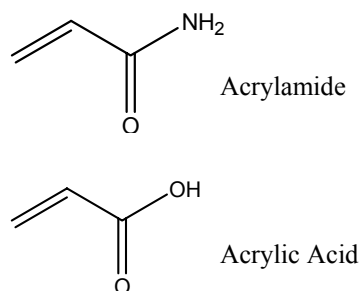


Figure 1: Structures of acrylamide and acrylic acid

### Introduction

Small polar molecules such as acrylamide and acrylic acid, (see figure 1), can be difficult if not impossible to separate on conventional reversed-phase bonded silica type phases using HPLC. Acrylamide historically has been analyzed using sample derivitization coupled with gas chromatography (1). The distinctive selectivity and increased retentiveness of the ZirChrom<sup>®</sup>-CARB phase, compared to bonded silica or polymeric phases, enables the baseline resolution of these small polar molecules.

### Experimental

A mixture of acrylamide monomer and acrylic acid was separated using ZirChrom<sup>®</sup>-CARB.

Column: ZirChrom<sup>®</sup>-CARB 50 mm x 4.6 mm i.d.  
Column Part #: ZR01-0546  
Mobile Phase: Gradient elution  
A: Acetonitrile  
B: 25mM phosphoric acid, pH 1.9

Time	% A	%B
0	0	100
4	25	75

Injection Vol.: 1 µl  
Pressure Drop: 180 bar  
Detection: UV at 210 nm  
Temperature: 40 °C with Metalox<sup>®</sup> 200-C  
Flow Rate: 2 ml/min

This method is an excellent example of how the enlarged zirconia method development “tool box” can be used to overcome the toughest separation challenges. Using temperature, extreme pH and the unique surface chemistry of ZirChrom<sup>®</sup>-CARB a fast separation of acrylamide and acrylic acid was achieved using a 4 minute gradient (see figure 2).

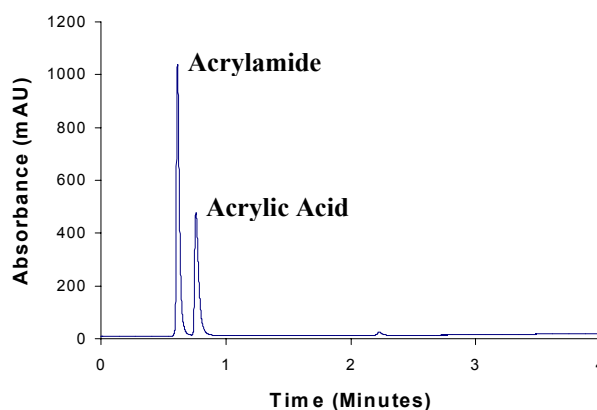


Figure 2: Separation of acrylamide and acrylic acid at 40 °C using a ZirChrom<sup>®</sup>-CARB column.

This method can be tailored to your specific application needs. ZirChrom method developers can help you optimize and transfer this method to your site. Please contact ZirChrom technical support at 1-866-STABLE1 or [support@zirchrom.com](mailto:support@zirchrom.com) for details.

ZirChrom phases offer unique selectivity for ionic compounds, high efficiency, and excellent chemical and thermal stability.

### References

- (1) E. Tareke, et al., “Acrylamide: A Cooking Carcinogen?” Chem. Red. Toxicol. 2000, 12, 517-522

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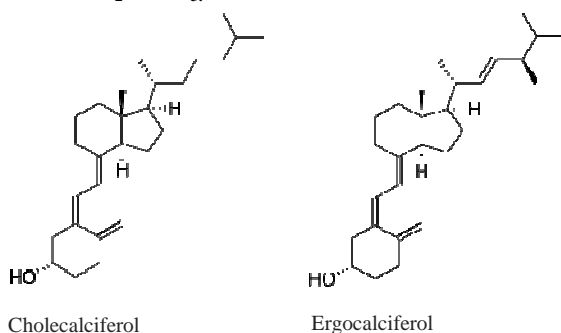
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# Fast Resolution of Vitamin D<sub>2</sub> from Vitamin D<sub>3</sub> on ZirChrom®-CARB

Merlin Bicking, Ph.D. and Kelly S. Johnson  
ACCTA and ZirChrom Separations, Inc.

## Technical Bulletin # 338

In this application we examine the superior selectivity of the ZirChrom®-CARB phase for two closely related compounds; vitamin D<sub>2</sub> and D<sub>3</sub>.



**Figure 1:** Structures of cholecalciferol (vitamin D<sub>3</sub>) and ergocalciferol (vitamin D<sub>2</sub>).

### Introduction

Vitamin D refers to a group of fat-soluble secosteroids which are critical for enhancing intestinal absorption of important nutrients. The most biologically relevant members of this group for humans are ergocalciferol (vitamin D<sub>2</sub>) and cholecalciferol (vitamin D<sub>3</sub>).

These closely related compounds are difficult to separate quickly using traditional reversed phase columns. Additionally, these highly reactive compounds are difficult to quantify precisely using LC/MS as the ionization process causing instability, lowering the robustness of the method (1).

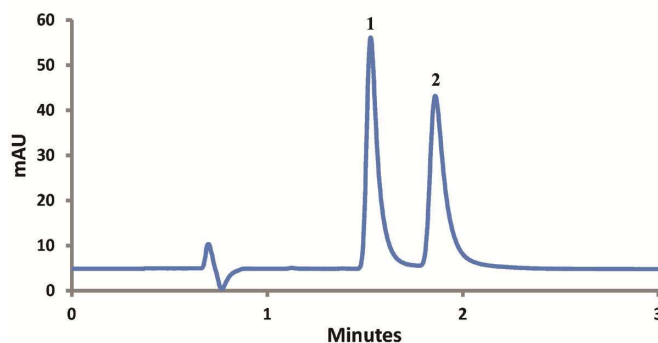
Here we present an isocratic method that provides baseline resolution of a mix of vitamin D<sub>2</sub> and D<sub>3</sub> standards in two minutes using a ZirChrom®-CARB phase with UV detection at 275 nm.

### Experimental

A mixture of two standards, cholecalciferol and ergocalciferol, was separated at 70 °C temperature using a ZirChrom®-CARB column. The separation conditions were as follows:

Column: ZirChrom®-CARB, 50 mm x 4.6 mm i.d.  
(Part Number: ZR01-5046)  
Mobile Phase: A: 50/50 ACN/IPA  
B: THF  
Temperature: 70 °C  
Flow Rate: 1.5 ml/min.  
Injection Vol.: 5 µl  
Pressure Drop: 74 bar  
Detection: UV at 275 nm

ZirChrom®-CARB separation of cholecalciferol and ergocalciferol allows for baseline resolution of the compounds in two minutes using isocratic conditions and UV detection.



**Figure 2:** 1 = Cholecalciferol (vitamin D<sub>3</sub>), 2 = Ergocalciferol (vitamin D<sub>2</sub>)

This method can be tailored to your specific application needs. ZirChrom technical support can help to optimize and transfer this method to your site. Please contact ZirChrom technical support at 1-866-STABLE-1 or [support@zirchrom.com](mailto:support@zirchrom.com) for details.

ZirChrom phases offer unique selectivity, high efficiency, and excellent chemical and thermal stability.

### References

(1) Brydwell, C.W. et al, Am J Clin Nutr, Vol. 88 no.2, 5545-5575 (2008)

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